

Fuel Cell Cathode with Liquid Water

Introduction

In low-temperature fuel cells, the produced water can condensate as liquid water if the partial pressure of vapor exceeds the (equilibrium) vapor pressure. The produced liquid water can flood the gas diffusion electrodes, gas diffusion layers, and/or the flow channels and manifolds, resulting in decreased fuel cell performance.

This tutorial expands the model defined in the Mass Transport and Electrochemical Reaction in a Fuel Cell Cathode tutorial to include also the effects liquid water formation in the porous gas diffusion electrode.

Model Definition

Liquid water is produced in the extended model using a user-defined expression for vapor condensation, depending on the relative humidity level in the gas phase:

$$R = k(p_{\rm H2O} - p_{\rm vap}) \tag{1}$$

where k is a rate constant, p_{H2O} is the partial pressure of water vapor, and p_{vap} the vapor pressure.

The porous gas diffusion electrode is assumed to be hydrophobic. The capillary pressure, p_c , is defined as

$$p_c = p_l - p_g \tag{2}$$

where p_l and p_g are the phase pressures, with the subscripts l and g referring to the liquid and gas phases, respectively.

The capillary pressure depends on the liquid saturation s_l in the porous media as depicted in Figure 1. (Note: The capillary pressure curve originally stems from Ref. 1 for measurements on a gas diffusion layer, that is, not a gas diffusion electrode, and is used here solely for tutorial purposes.)



Figure 1: Capillary pressure versus liquid saturation.

The momentum transfer in the liquid phase is modeled using Darcy's law according to

$$\nabla \cdot (\rho_l \mathbf{u}_l) = R \tag{3}$$

with the volume averaged velocity \mathbf{u}_i in the liquid fluid phase of index l defined as

$$\mathbf{u}_{l} = -\frac{\kappa_{r,l}}{\mu_{l}} \kappa \nabla p_{l} \tag{4}$$

where μ_l is the dynamic viscosity of the fluid and κ the absolute permeability of the porous media.

 $\kappa_{r,l}$ in the above equation is the relative permeability, which is defined to depend on the fluid saturation level according to

$$\kappa_{r,l} = s_l^2 \tag{5}$$

A similar correlation is used for Darcy's law in the gas phase, solved for by the fuel cell interface, i.e

$$\kappa_{r,g} = s_g^2 \tag{6}$$

The Phase Transport in Porous Media interface is used to set up the flow formulation of the liquid water, based on Darcy's law, solving for s_l , to the model.

As boundary conditions for the liquid phase transport model, a liquid water saturation according to a capillary pressure of 0 Pa is set at the inlet.

The model is solved in a stationary study consisting of four study steps. In order to facilitate convergence, the potential variables, the pressure and gas phase species, and the liquid saturation are solved for in individual study steps first for a cell voltage of 1 V. The fully coupled problem is then solved in the last study step, using an auxiliary sweep to ramp the voltage from 1 to 0.5 V.

Results and Discussion

Figure 2 compares the relative humidity level in the cell at 0.5 V when and when not including water condensation and liquid water transport in the model. When not including water condensation, the air mixture gets significantly oversaturated to about 160% relative humidity. When including condensation, the relative humidity stays close to 100% throughout the cathode.



Figure 2: Relative humidity assuming no water condensation (left) and when including condensation and water transport (right).

Figure 3 shows the liquid saturation level and the corresponding capillary pressure level when including liquid water transport. Liquid water now forms in the electrode and is

transported out toward the inlet hole mainly by the capillary pressure gradient. The water saturation level in the pores does not exceed 10%.



Figure 3: Liquid water saturation level and liquid water flux (left) and corresponding capillary pressure distribution (right).

Finally, the polarization curves for the two cases are compared in Figure 4. Introducing liquid water transport in the model slightly increases the current levels for a given voltage especially for lower voltages. This may seem counterintuitive, but the reason for this is that the condensation of water vapor results in an increased partial pressure of oxygen, which has a positive effect on the cathode potential. The small volume (<10%) of liquid water has only a minor detrimental effect on the overall gas transport rate.



Figure 4: Comparisons of polarization plots when, and when not, considering liquid water formation and transport in the cathode.

Reference

1. E.C. Kumbur, K.V. Sharp, and M.M. Mench, "Validated Leverett Approach for Multiphase Flow in PEFC Diffusion Media, I. Hydrophobicity Effect," *Journal of The Electrochemical Society*, vol. 154, no. 12, pp. B1295–B1304, 2007.

Application Library path: Fuel_Cell_and_Electrolyzer_Module/Fuel_Cells/ fuel_cell_cathode_with_liquid_water

Modeling Instructions

APPLICATION LIBRARIES

I From the File menu, choose Application Libraries.

- 2 In the Application Libraries window, select Fuel Cell and Electrolyzer Module>Fuel Cells> fuel_cell_cathode in the tree.
- 3 Click 🔮 Open.

First, make a simulation for an inlet relative humidity of 100%.

GLOBAL DEFINITIONS

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** In the table, enter the following settings:

Name	Expression	Value	Description
RH	100[%]	1	Relative humidity

STUDY I

In the **Home** toolbar, click **= Compute**.

RESULTS

Make a plot of the relative humidity as follows:

Relative Humidity

- I In the Home toolbar, click 🚛 Add Plot Group and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Relative Humidity in the Label text field.

Volume 1

- I Right-click Relative Humidity and choose Volume.
- In the Settings window for Volume, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>
 Hydrogen Fuel Cell>fc.RH Relative humidity.

Selection 1

- I Right-click Volume I and choose Selection.
- 2 In the Settings window for Selection, locate the Selection section.
- **3** From the Selection list, choose Cathode Gas Diffusion Electrode.

4 In the **Relative Humidity** toolbar, click **I** Plot.



Probe Table 1

Store the probe table containing the polarization data for later comparisons.

Oversaturated, no liquid water transport

- I In the Model Builder window, expand the Results>Tables node.
- 2 Right-click Probe Table I and choose Duplicate.
- **3** In the **Settings** window for **Table**, type **Oversaturated**, no liquid water transport in the **Label** text field.

COMPONENT I (COMPI)

Now set up the liquid transport model.

ADD PHYSICS

- I In the Home toolbar, click 🙀 Add Physics to open the Add Physics window.
- 2 Go to the Add Physics window.
- 3 In the tree, select Fluid Flow>Multiphase Flow>Phase Transport> Phase Transport in Porous Media (phtr).
- **4** Click to expand the **Dependent Variables** section. In the **Volume fractions** table, enter the following settings:

s_g

s_l

- 5 Click Add to Component I in the window toolbar.
- 6 In the Home toolbar, click 🙀 Add Physics to close the Add Physics window.

GLOBAL DEFINITIONS

The gas volume fraction will be a variable in the modified model, depending on the liquid saturation level. Locate and change the eps_gas parameter name and description as follows:

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** In the table, enter the following settings:

Name	Expression	Value	Description
eps_pores	0.4	0.4	Pore volume fraction in porous electrode

Also, add a parameter kce to define the condensation evaporation rate constant.

4 In the table, enter the following settings:

Name	Expression	Value	Description
kce	5.62e4[mol/m^3/s]	56200 mol/(m ³ ·s)	Condensation evaporation rate constant

DEFINITIONS

Load some variable expressions from a file:

Variables I

- I In the Model Builder window, under Component I (compl) right-click Definitions and choose Variables.
- 2 In the Settings window for Variables, locate the Variables section.
- 3 Click 📂 Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file fuel_cell_cathode_with_liquid_water_variables.txt.

The pc variable will be marked in orange since the capillary pressure function has not yet been defined. Define the function as follows:

Interpolation 1 (int1)

- I In the Home toolbar, click f(X) Functions and choose Local>Interpolation.
- 2 In the Settings window for Interpolation, locate the Definition section.
- 3 From the Data source list, choose File.

- 4 Click **Browse**.
- 5 Browse to the model's Application Libraries folder and double-click the file fuel_cell_cathode_with_liquid_water_pc.txt.
- 6 Click **[III]** Import.
- 7 In the Function name text field, type pc.
- 8 Locate the Interpolation and Extrapolation section. From the Extrapolation list, choose Linear.
- 9 Locate the Units section. In the Argument table, enter the following settings:

Argument	Unit
t	1

IO In the **Function** table, enter the following settings:

Function	Unit
рс	bar

II Click to expand the **Related Functions** section. Select the **Define inverse function** check box.

12 In the **Inverse function name** text field, type **s_1**.

I3 Click 💿 Plot.



Variables I

Now go back and check so that there are no variable marked in orange in the list.

If there are variables marked in orange it could be that you missed setting the dependent variables to s_g and s_1 when adding **Phase Transport In Porous Media**, or that you missed setting the Function name to pc when adding the **Interpolation** function, or missed renaming eps gas to eps pores in **Parameters**.

HYDROGEN FUEL CELL (FC)

Enable liquid water stoichiometry as follows:

- I In the Model Builder window, expand the Component I (compl)>Hydrogen Fuel Cell (fc) node, then click Hydrogen Fuel Cell (fc).
- 2 In the Settings window for Hydrogen Fuel Cell, locate the O2 Gas Mixture section.
- **3** Find the **Reactions** subsection. Select the **Include H2O(I)** in reaction stoichiometry check box.

O2 Gas Phase 1 Add the condensation reaction as follows: In the Model Builder window, under Component I (comp1)>Hydrogen Fuel Cell (fc) click O2 Gas Phase I.

Water Condensation-Evaporation I

- I In the Physics toolbar, click 📃 Attributes and choose Water Condensation-Evaporation.
- 2 In the Settings window for Water Condensation-Evaporation, locate the Condensation-Evaporation Rate section.
- **3** In the k_{ce} text field, type kce.

O2 Gas Diffusion Electrode I

Due to the presence of liquid water in the cathode, update the permeability as follows:

- I In the Model Builder window, under Component I (compl)>Hydrogen Fuel Cell (fc) click O2 Gas Diffusion Electrode I.
- 2 In the Settings window for O2 Gas Diffusion Electrode, locate the Gas Transport section.
- 3 In the κ_g text field, type perm_eff_gas.

PHASE TRANSPORT IN POROUS MEDIA (PHTR)

Now set up the phase transport as follows:

- I In the Model Builder window, under Component I (compl) click Phase Transport in Porous Media (phtr).
- **2** In the Settings window for Phase Transport in Porous Media, locate the Domain Selection section.
- 3 From the Selection list, choose Cathode Gas Diffusion Electrode.

Phase and Porous Media Transport Properties I

- In the Model Builder window, under Component I (compl)>
 Phase Transport in Porous Media (phtr) click
 Phase and Porous Media Transport Properties I.
- 2 In the Settings window for Phase and Porous Media Transport Properties, locate the Model Input section.
- **3** In the p_A text field, type fc.pA.
- **4** Locate the **Capillary Pressure** section. In the p_{csl} text field, type pc.

The Hydrogen Fuel Cell interface declares and announces density and viscosity variables for both the gas mixture and liquid water. Define the corresponding properties in the Phase Transport interface to make use of these variables as follows:

5 Locate the Phase I Properties section. From the ρ_{sg} list, choose Density of gas phase (fc).

- 6 From the μ_{sg} list, choose Dynamic viscosity of gas phase (fc).
- 7 Locate the Phase 2 Properties section. From the ρ_{sl} list, choose Density of liquid water (fc).
- 8 From the μ_{sl} list, choose Dynamic viscosity of liquid water (fc).
- 9 Locate the Matrix Properties section. From the ϵ_p list, choose User defined. In the associated text field, type eps_gas.
- IO From the κ list, choose User defined. In the associated text field, type perm.

Volume Fraction 1

I In the Physics toolbar, click 🔚 Boundaries and choose Volume Fraction.

At the inlet boundary, the liquid saturation corresponds to a zero capillary pressure.

- 2 In the Settings window for Volume Fraction, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Inlet**.
- 4 Locate the Volume Fraction section. Select the Phase s_I check box.
- **5** In the $s_{0,\text{sl}}$ text field, type $s_1(0)$.

If s_1(0) gets marked in orange you probably missed defining the inverse function on the **Interpolation** function.

Initial Values 1

Use the same expression for the initial volume fraction.

- I In the Model Builder window, click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the $s_{0,sl}$ text field, type $s_1(0)$.

Mass Source I

Add the mass source for the liquid phase, stemming from the fuel cell reactions, as follows:

- I In the Physics toolbar, click ⊨ Domains and choose Mass Source.
- 2 In the Settings window for Mass Source, locate the Domain Selection section.
- 3 From the Selection list, choose All domains.
- 4 Locate the Mass Source section. From the $qs_{\rm sl}$ list, choose Mass source, liquid phase (fc/ o2gasph1).

STUDY I

Use a study sequence consisting of four steps to solve the model. The stepwise approach improves convergence. Modify the existing study as follows:

Stationary - Excluding Phase Transport

- I In the Study toolbar, click 🦰 Study Steps and choose Stationary>Stationary.
- 2 In the Settings window for Stationary, type Stationary Excluding Phase Transport in the Label text field.
- **3** Locate the **Physics and Variables Selection** section. In the table, clear the **Solve for** check box for **Phase Transport in Porous Media (phtr)**.

Step 2: Stationary - Excluding Phase Transport

Right-click Study I>Step 3: Stationary - Excluding Phase Transport and choose Move Up.

Stationary - Phase Transport Only

- I In the Study toolbar, click *Study Steps* and choose Stationary>Stationary.
- 2 In the Settings window for Stationary, type Stationary Phase Transport Only in the Label text field.
- **3** Locate the **Physics and Variables Selection** section. In the table, clear the **Solve for** check box for **Hydrogen Fuel Cell (fc)**.

Step 3: Stationary - Phase Transport Only

Right-click Study I>Step 4: Stationary - Phase Transport Only and choose Move Up.

Stationary - All Physics

- I In the Model Builder window, click Step 4: Stationary.
- 2 In the Settings window for Stationary, type Stationary All Physics in the Label text field.

The study should now contain four steps, in the following order: Step 1: Current Distribution Initialization, Step 2: Stationary - Excluding Phase Transport, Step 3: Stationary - Phase Transport Only, Step 4: Stationary - All Physics.

Solution 1 (soll)

- I In the Model Builder window, right-click Solver Configurations and choose Reset Solver to Default.
- 2 In the Study toolbar, click **=** Compute.

RESULTS

Inspect the relative humidity plot.

Relative Humidity

In the **Relative Humidity** toolbar, click **O** Plot.



Liquid Water Saturation

Create a plot of the liquid saturation level as follows:

- I In the Home toolbar, click 📠 Add Plot Group and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Liquid Water Saturation in the Label text field.

Volume 1

- I Right-click Liquid Water Saturation and choose Volume.
- 2 In the Settings window for Volume, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>
 Phase Transport in Porous Media>s I Volume fraction.
- **3** In the Liquid Water Saturation toolbar, click **O** Plot.

Arrow Surface 1

- I In the Model Builder window, right-click Liquid Water Saturation and choose Arrow Surface.
- In the Settings window for Arrow Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>
 Phase Transport in Porous Media>phtr.ux_s_l,...,phtr.uz_s_l Volumetric flux, phase s_l.
- 3 Locate the Arrow Positioning section. In the Number of arrows text field, type 50.
- 4 Locate the Coloring and Style section. From the Color list, choose Black.

Selection 1

- I Right-click Arrow Surface I and choose Selection.
- **2** Select Boundary 10 only.

3 In the Liquid Water Saturation toolbar, click **9** Plot.



Capillary Pressure

Create a plot of the capillary pressure as follows:

- I In the Home toolbar, click 🚛 Add Plot Group and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Capillary Pressure in the Label text field.

Volume 1

- I Right-click Capillary Pressure and choose Volume.
- 2 In the Settings window for Volume, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>Definitions> Variables>pc Capillary pressure Pa.
- 3 Locate the Expression section. From the Unit list, choose mbar.
- **4** In the **Capillary Pressure** toolbar, click **OD Plot**.





Polarization Curve

Finally, compare the polarization plots of the oversaturated and liquid water models as follows:

Probe Table Graph: Limited O2 gas phase transport

- I In the Model Builder window, expand the Polarization Curve node, then click Probe Table Graph: Limited O2 gas phase transport.
- 2 In the Settings window for Table Graph, click to expand the Legends section.
- **3** In the table, enter the following settings:

Legends

With liquid water

Probe Table Graph: Unlimited O2 gas phase transport

- I In the Model Builder window, click Probe Table Graph: Unlimited 02 gas phase transport.
- 2 In the Settings window for Table Graph, locate the Data section.
- **3** From the Table list, choose **Oversaturated**, no liquid water transport.
- **4** Locate the **Legends** section. In the table, enter the following settings:

Legends

Oversaturated, no liquid water

5 In the Polarization Curve toolbar, click 💿 Plot.



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